## A Kinetic Study of the Densification of TiB<sub>2</sub> at High Pressure and High Temperature

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High-pressure technology has been used to prepare dense, fine-grain specimens of  $TiB_2$  at temperatures as low as 1600°C, as compared to temperatures of the order of 2300°C which are required to prepare dense, coarse-grain specimens by vacuum-sintering. Experiments were performed to describe the temperature and the pressure coefficients of the rate of densification and the effect of temperature and pressure on grain growth. Fully dense specimens of  $TiB_2$  can be fabricated without significant grain growth. A mechanism for densification under these extreme experimental conditions is suggested.

## INTRODUCTION

Several fabricating procedures were studied in the initial phase of a fundamental investigation [<sup>1</sup>] of the physical, chemical, and mechanical properties of the diborides of titanium, zirconium, hafnium, niobium, and tantalum. During the early stages of this program, it became apparent that neither vacuum- nor inert-atmosphere sintering [<sup>2</sup>] nor conventional hot-pressing procedures could provide suitable specimens for the above program. The floating-zone-melting technique [<sup>3</sup>] was used to prepare pure samples of all the above diborides except TiB<sub>2</sub>, the lack of mechanical integrity of zone-refined NbB<sub>2</sub> and TaB<sub>2</sub> precluded the use of these materials for physical and mechanical property measurements. Concurrently, a limited study was initiated to ascertain the feasibility of preparing material suitable for various property measurements by using high-pressure technology to extend the pressure range of conventional hot-pressing procedures to the range 100,000–300,000 psi. The successful application of this technique was demonstrated by the fabrication of specimens for oxidation evaluation and for physical and mechanical property measurements [<sup>4</sup>].

The present program was undertaken to study the mechanism of densification of  $TiB_2$  at high pressures and high temperatures;  $TiB_2$  was chosen as representative of the hard-metal compounds, which are difficult to fabricate without the use of low-melting binder materials.

## EXPERIMENTAL

## **Characterization of Starting Material**

The  $TiB_2$  powder was a high-purity product supplied by Millmaster Chemical Company. The chemical and spectroscopic analysis data given in Table I indicate